

Title: Focus SPM Chemistry 2023

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Errata

Page number <i>Halaman</i>	Section/ Part <i>Bahagian</i>	Error <i>Kesalahan</i>	Correction <i>Pembetulan</i>
126	What is a Base? 1.	Arrhenius' definition of a base: A base is a chemical substance that ionises in water to produce hydroxide ions, OH ⁻ .	Definition of base: A base is a chemical substance that reacts with acids to produce salt and water.
243	Example 1.1 (Solution)	Ammonia is an oxidising agent because it loses hydrogen in the reaction. Copper(II) oxide is a reducing agent because it loses oxygen in the reaction.	Ammonia is a reducing agent because it loses hydrogen in the reaction. Copper(II) oxide is an oxidising agent because it loses oxygen in the reaction.
283	Activity 1.6 Table of half- cell standard electrodes potential (anode)	Column: 1, Row: 1 $O_2 + 2H_2O + 4e^- \rightarrow 4OH^-$	Column: 1, Row: 1 $O_2 + 2H_2O + 4e^- \rightarrow 4OH^-$
498	Answer Checkpoint 3.1	Q1. 16	Q1. 12
508	ANSWER FORM 5 SPM Practice 1	3. B	3. C